

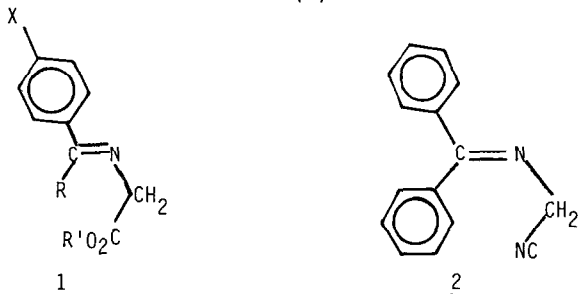
SYNTHESIS OF  $\alpha$ -AMINOACIDS BY CATALYTIC  
 PALLADIUM (0) ALKYLATION OF SCHIFF BASES

D.FERROUD ; J.P.GENET\* AND R.KIOLLE (in part)

Laboratoire de Synthèse Organique et Organométallique, Laboratoire  
 Associé au C.N.R.S., Université Pierre et Marie CURIE, 8 Rue Cuvier 75005 PARIS

*Abstract* : Schiff bases 1 b, 1 c, 2 derived from glycine ester or aminoacetonitrile were alkylated with allylic acetates 3 a, 3 b or allylic carbonates 3 c, 3 d, 4 a, 4 b (under neutral conditions) in the presence of catalytic amount of palladium (0). After hydrolysis higher and functionalized  $\alpha$ -aminoesters were obtained in good yields (50 to 85 %).

Since the first report of STORK *et al* (1) on the alkylation of the anion of Schiff base 1 a derived from glycine ester, efficient and elegant routes to  $\alpha$ -aminoacids have been widely developed. Schiff bases can be alkylated in the presence of strong bases (2) and under various phase transfer conditions (3).



- 1  
2
- a : R=X=H ; R=C<sub>2</sub>H<sub>5</sub>  
 b : R=H ; X=Cl ; R'=C<sub>2</sub>H<sub>5</sub>  
 c : R=Ph ; X=H ; R'=CH<sub>3</sub>

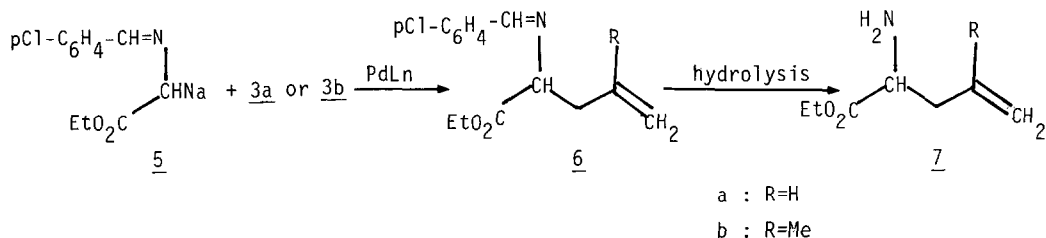
As part of our program on the  $\alpha$ -aminoacids synthesis we investigated very recently the palladium catalyzed alkylation of nitroacetic esters (4,5) and Schiff bases derived from  $\alpha$ -aminoesters. We wish to report here a convenient approach for the synthesis of higher and functionalized  $\alpha$ -aminoacids by palladium (0) alkylation of Schiff bases (6) with various allylic derivatives 3, 4 ; (Table I).



- 3  
4
- a : R'=H ; R=COCH<sub>3</sub>  
 b : R'=CH<sub>3</sub> ; R=COCH<sub>3</sub>  
 c : R'=H ; R=CO<sub>2</sub>C<sub>2</sub>H<sub>5</sub>  
 d : R'=CH<sub>3</sub> ; R=CO<sub>2</sub>C<sub>2</sub>H<sub>5</sub>
- a : R=Ph  
 b : R=CH<sub>2</sub>OAc

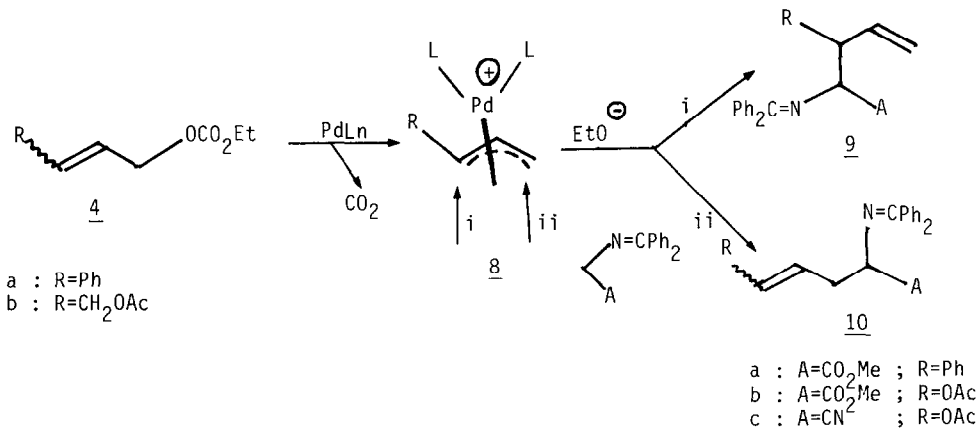
Schiff bases of glycine ester or aminoacetonitrile derived from aromatic aldehydes or ketones (e.g. p-chlorobenzaldehyde and benzophenone) have been chosen because they are crystalline starting materials readily available (7) instead of Schiff base 1<sub>a</sub> obtained from benzaldehyde which is a less stable oil.

The catalytic classical palladium-alkylation of 1<sub>b</sub>, with allylic acetates 3<sub>a</sub>, 3<sub>b</sub>, was carried out with the preformed anion of Schiff base (from NaH in THF) and gave after hydrolysis and extractive work up higher monoalkylated  $\alpha$ -aminoesters 7 in 50-65 % yield (runs 1,2).



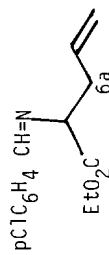
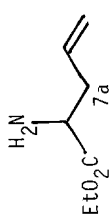
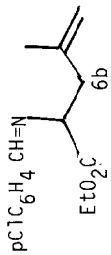
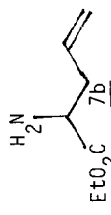

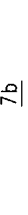
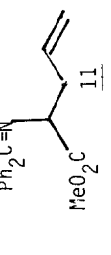
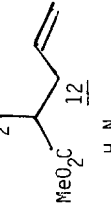
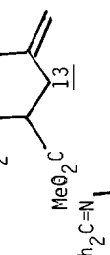
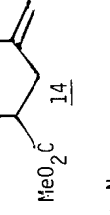
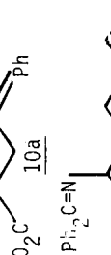
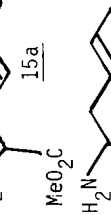
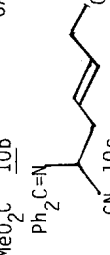
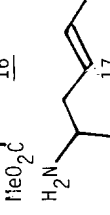
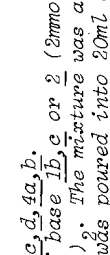
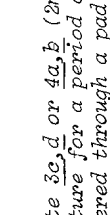
More interestingly the palladium-catalyzed alkylation can be performed *under neutral conditions* starting from allylcarbonates 3<sub>c</sub>, 3<sub>d</sub>, 4<sub>a</sub>, 4<sub>b</sub> (8). This alkylation without preformation of the anion of Schiff bases gave only monoalkylated products 6<sub>b</sub>, 11, 13 and after hydrolysis the  $\alpha$ -aminoesters 7<sub>b</sub>, 12, 14, in good yield (70-90%) (runs 3,4,5) (Table I).

The regioselectivity of the alkylation of 1<sub>c</sub> and 2 was tested with carbonate of cinnamyl alcohol 4<sub>a</sub> and with the bis allylic - 1,4 acetate carbonate 4<sub>b</sub>.



The reaction of 4<sub>a</sub> occurs in good yield (80%) almost completely at the less substituted end (path ii) of  $\eta^3$  palladium complex 8 with formation of 9<sub>a</sub> and 10<sub>a</sub> (9) in a ratio 1/19 (run 6). In contrast, alkylation of 4<sub>b</sub> was completely regioselective and we obtained mono-functionalized alkylated 10<sub>b</sub>, 10<sub>c</sub> products (10) in good yields (70-80%) (runs 7,8) (Table I).

TABLE I

Run	Schiff base	Allylic substrate	Conditions <i>t</i> (h), <i>T</i> (°C)	Products	(yield %)	Aminester (b)	(yield %)
1	<u>1b</u>	<u>3a</u>	1 20	 <u>6a</u>	70	 <u>7a</u>	50
2	<u>1b</u>	<u>3b</u>	1 20	 <u>6b</u>	60	 <u>7b</u>	65
3	<u>1b</u>	<u>3d</u>	1 20	 <u>6b</u>	70	 <u>7b</u>	65
4	<u>1c</u>	<u>3c</u>	2 20	 <u>11</u>	80	 <u>12</u>	75
5	<u>1c</u>	<u>3d</u>	2 20	 <u>13</u>	90	 <u>14</u>	85
6	<u>1c</u>	<u>4a</u>	8 20	 <u>10a</u>	80	 <u>15a</u>	70
7	<u>1c</u>	<u>4b</u>	8 20	 <u>10b</u>	70	 <u>16</u>	75
8	<u>2</u>	<u>4b</u>	8 20	 <u>10c</u>	80	 <u>17</u>	70

a) General Procedure with allylic carbonates 3c, 3d, 4a, 4b.

Under argon to a stirred solution of Schiff base 1b, 1c or 2 (2mmoles) and allyl carbonate 3c, 3d or 4a, 4b (2mmoles) in 2ml of dry THF was added 0.025 to 0.05 mmol of Pd(dppe)<sub>2</sub>. The mixture was allowed at room temperature for a period of extended time. The solvent was evaporated, the crude material was poured into 20ml of ether and then filtered through a pad of Celite. After removal of solvent, the mixture was chromatographed on silica gel to give pure alkylated Schiff bases.

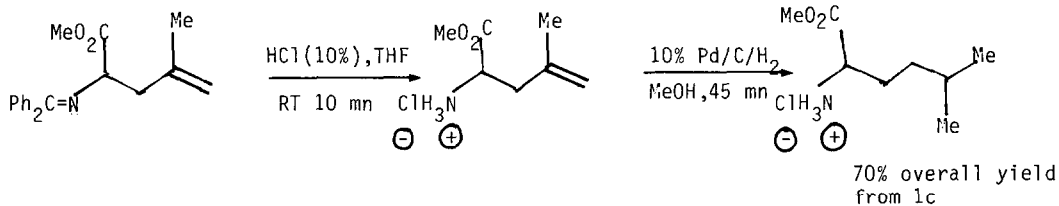
b) After hydrolysis with HCl (10%) at room temperature for 10min and classical work up.

The palladium catalyzed alkylation under neutral conditions provides a particularly attractive approach for the preparation of  $\beta,\gamma$ -unsaturated  $\alpha$ -aminoacids derivatives and natural  $\alpha$ -aminoacids with interesting biological properties. For example, in this paper we synthesized the methyl ester of trans-2, amino-5, phenyl-4, pentenoic acid 15a inhibitor of S-adenosyl transferase (11) (under its acid form); allyl glycine methyl ester and leucine methyl ester (12) were also prepared (13).

The applicability of this process to asymmetric synthesis is under investigation.

#### REFERENCES

- 1) G.Stork, A.Y.W Leong and A.M. Touzin, *J.Org.Chem.* **41**, 3491 (1976).
- 2) a) P.Bey, J.P.Ververt, V.Van.Dorsselaer and M.Kolb, *J.Org.Chem.* **44**, 2732 (1979) and references therein. b) T.Yamashita, H.Mitsui, H.Watanabe and N.Nakamura, *Bull.Chem.Soc. Japan* **55**, 961 (1982) and references therein. c) N.Minowa, S.Fukatu, T.Niida, M.Takada, K.Sato. *Tetrahedron Lett.* **24**, 2391 (1983). d) P.Duhamel, J.J.Eddine and J.Y.Valnot, *Tetrahedron Lett.* **25**, 2355 (1984).
- 3) M.J.O'Donnell, B.Leclef and D.B.Rusterholz, L.Ghosez, J.P.Antoine and M.Navarro, *Tetrahedron Lett.* **23**, 4259 (1982); *Synthesis*, 313, (1984).
- 4) J.P.Genêt and D.Ferroud, *Tetrahedron Lett.* **25**, 3579 (1984).
- 5) D.Ferroud, J.P.Genêt and J.Muzart, *Tetrahedron Lett.* **25**, 4379 (1984).
- 6) To our knowledge only one example of catalytic alkylation of Schiff base was reported. J.P.Haudegond, D.Commereuc, J.Collin, and Y.Chauvin *C.A.96*, P 85976 c, FR2, 474,491.
- 7) M.J.O'Donnell and R.L.Polt, *J.Org.Chem.* **47**, 2663 (1982).
- 8) J.Tsuiji, I.Minami and I.Shimizu, *Tetrahedron Lett.* **24**, 5635 (1983) and references therein.
- 9) The two regioisomers were purified by flash chromatography over alumina, eluent : 10 % Ethyl acetate in Hexane.  
10a : IR (film) 3030, 2950, 1730, 1620, 1435, 1250, 1170, 970  $\text{cm}^{-1}$ . RMN  $^1\text{H}$  80 MHz  $\text{CDCl}_3$ : 7.75-6.95 (m 15H); 6.5-5.8 (m 2H); 4.20 (t 1H); 3.75 (s 3H); 2.82 (m 2H).  
9a : IR (film) 3030, 3020, 2910, 1730, 1620, 1595, 1570, 1440, 1310, 960  $\text{cm}^{-1}$ .  
 RMN (80 MHz,  $\text{CDCl}_3$ ) 8-7.15 (m 15H); 6.75-6 (m 1H); 5.45-5.12 (m 2H); 4.25 (t 1H); 4.12 (m 1H); 3.75 (s 3H).
- 10) 10c : IR (film) 3020, 2950, 2230, 1730, 1610, 1440, 1360, 1290, 1230, 1020, 960. RMN 80 MHz  $\text{CDCl}_3$  : 7.87-7.15 (m 10H); 5.75 (m 2H); 4.5 (m 2H); 4.3 (t 1H); 2.67 (m 2H); 2.02 (s 3H).  
10b : RMN (60MHz,  $\text{CDCl}_3$ ) : 7.75-6.95 (m 10H); 5.50 (m 2H); 4.42 (m 2H); 4.10 (t 1H) 3.75 (s 3H); 2.6 (m 2H); 2.00 (s 3H).  
 IR (film) 2950, 2920, 1720, 1590, 1450, 1365, 1230, 1010, 960.
- 11) A.W.Coulter, J.B.Lombardini and P.Talalay, *Mol.Pharm.* **10**, 305 (1974).
- 12)



- 13) We wish to thank Centre National de la Recherche Scientifique (CNRS) for financial support : D.F.thanks DGRST for Grant (1983-1985).

(Received in France 25 October 1985)